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TiO$_2$-PORPHYRIN NANOSTRUCTURES FOR AMINO ACID DETECTION

A novel optical sensor based on TiO$_2$ nanoparticles for Valine (one of the twenty standard amino acids within proteins) detection has been developed. In the presented work, commercial TiO$_2$ nanoparticles (Sigma Aldrich, particle size 32 nm) were used as sensor templates. The sensitive layer was formed by a porphyrin coating on a TiO2 nanostructured surface. As a result, an amorphous layer between the TiO$_2$ nanostructure and porphyrin was formed. Photoluminescence (PL) spectra were measured in the range of 370-900 nm before and after porphyrin application. Porphyrin adsorption led to a decrease of the main TiO$_2$ peak at 510 nm and the emergence of an additional peak of high intensity at 700 nm. Absorption spectra (optical density vs. wavelenght, measured from 300 to 800 nm) also showed great changes; absorption edge shift and additional peaks appearing. Adsorption of amino acid resulted in a decrease of the intensity of the PL peak due to porphyrin and an increase of intensity of the TiO$_2$ main PL peak. The interaction between the sensor surface and the amino acid leads to the formation of new complexes on the surface and results in a reduction of the optical activity of porphyrin. Sensitivity of the sensor with respect to different concentrations of Valine was calculated. The developed sensor can determine the consentration of Valine in the range of 0.04 to 0.16 mg/ml.

1. INTRODUCTION

Amino acids are complex molecules forming building blocks of proteins and involved in metabolism as intermediates. There are twenty amino acids involved in protein construction. Each of them contains a unique functional group, which defines the fundamental properties such as size, shape, charge, capacity for hydrogen bonding, hydrophilicity/hydrophobicity and chemical reactivity. Valine ($C_5H_{11}NO_2$), the one of the most important amino acids, is a branched-chain essential amino acid, hydrophobic and usually localized inside of proteins [1]. It is a stimulating agent which promotes muscle growth and tissue regeneration [2,3]. Valine can be used as food additive [4,5], nutrient and/or dietary supplement in animal drugs, feeds, and related products [6,7]. Because of the above mentioned properties, Valine is often used by bodybuilders (in conjunction with leucine and isoleucine) as stimulating

agent. However, high concentrations of Valine can induce a crawling sensation on the skin and hallucinations [8], what is crucial for people with kidney or liver disease. Therefore, the determination of the Valine concentration in human body is an important task in medicine.

Titanium dioxide is chemically stable, non-toxic and a low-cost material which is well known for its good optical, photocatalytic and sensing properties [9-15]. Over the last decade, TiO$_2$ nanostructures, due to quantum-size effects such as absorption edge shift and the appearance of photoluminescence at room temperature, have been increasingly used as a sensor platform [16-26]. Recently, there has been a growing interest in the development of a new class of hybrid systems - TiO$_2$ -sensitizers, in which macrocycles such as porphyrins are used to form the sensitive layer [24]. Porphyrins are brightly colored pigments, containing nitrogen, that consist of conjugated multiple-loop systems, based on six-

teen-membered microcycles, composed of four pyrrole molecules and bridges. A porphyrin molecule contains a coordination cavity, bound by four nitrogen atoms, having a radius of about 2Å. This molecule is capable to coordinate with metal ions of different degree of oxidation. As a result, porphyrin-metal complexes, so-called metalloporphyrins, are being formed, that possess unique combinations of structural, physical and chemical features with high biological and catalytic activity.

It is known that porphyrin application enhances photocatalytic activity of the samples. In [27], the role of both metal and macrocycle in the photocatalytic processes have been studied by utilizing TiO_2 samples coated by porphyrins and metalloporphyrins. Significant changes in optical properties of nanoporous glass filled with TiO_2 and TiO_2/porphyrin nanostructures have previously been found [28]. In this paper we report on the investigation of new optical biosensor based on TiO_2 nanoparticles coated by porhyrin for Valine detection.

2. EXPERIMENTAL

Commercial TiO_2 nanoparticles (Sigma Aldrich, particle size 32 nm) were used as a biosensor template. TiO_2 nanoparticles were dissolved in water to prepare sols. TiO_2 layers were formed on glass substrates by dropping TiO_2 sols on the substrate and drying it at room temperature [21]. Post annealing treatment at 300 0C for 1 hour was performed to remove water from the samples. Structural properties of the obtained samples were studied by SEM.

The fabrication of sensitive layers was performed by dropping of porphyrin "5,15-di(n-nonyl),10,20-di(4-pyridyl) porphynatotin dichloride" (chemical structure is shown in Figure 1) solution in chlorophorm on TiO_2 surface. Photoluminescence (PL) spectra were measured with the setup presented in Figure 2. The samples were excited by a UV laser (LCS-DTL-374QT, L_{ex}=355 nm) and PL spectra were recorded in the wavelength range of 370-800 nm. Absorbance spectra were measured with the use of a UV-VIS spectrophotometer (Shimadzu UV-1700) in the range of 350-1100 nm.

To check the sensitivity of porphyrin to Valine, PL spectra of porphyrin layer before and after interaction with Valine were studied. To study biosensor response, different concentrations of Valine in aqueous solution were deposited on TiO_2-porphyrin surfaces.

The spectra, measured after Valine deposition, showed no drastic changes in the PL intensity and peak position (see Fig.3 in sec.).
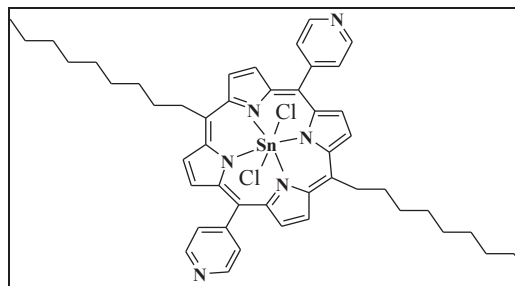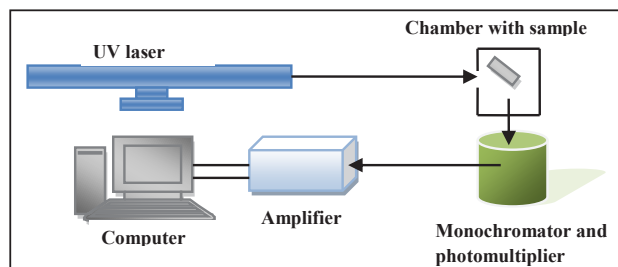

Fig.1. **Chemical structure of porphyrin**

Fig.2. **Experimental setup for photoluminescence measurements**

3. RESULTS AND DISCUSSION

The obtained TiO_2 nanostructures were rough and porous as it is shown in Figure 3. Absorption spectra of initial porphyrin layer and porphyrin coated TiO_2 nanostructure are shown on Figure 4. The porphyrin demonstrated a Sorret band absorption, centered at 424 nm. It was found that after deposition of porhyrin on TiO_2, the Sorret band was shifted toward IR region, matching the interaction TiO_2-porphyrin.

Deposition of porphyrin layer resulted in significant changes in the PL spectrum of TiO_2-porhyrin nanostructure (Figure 4). Initially, TiO_2 emission spectrum showed wide peak, centered at 510 nm and the poprhyrin emission was centered at 693 nm.
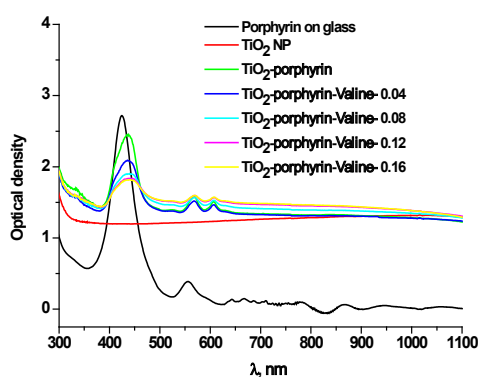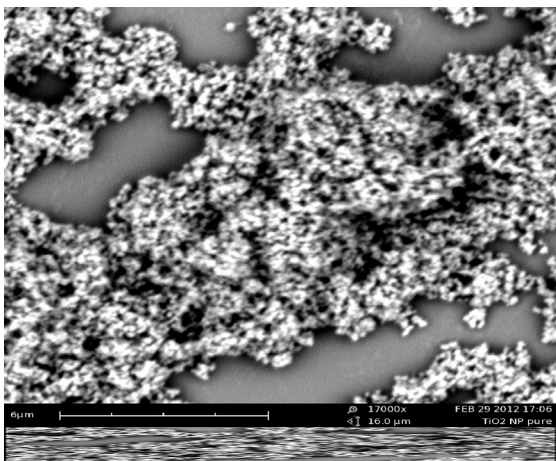
Fig.3. **SEM image of TiO$_2$ nanostructures**
Fig.4. **Absorption spectra of the studied samples**

The peak, related to pure TiO$_2$ was quenched by a factor of three, while a peak, related to porphyrin, shifted to 700 nm after the formanation of TiO$_2$-porphyrin complex (Fig.5). The obtained PL data confirm the absorption results, matching to the interaction between metal oxide and porphyrin (ADD pure porphyrin spectra in the PL to discuss the changes).

We suggest that the creation of the porphyrin-metal oxide structure was caused by the formation of an amorphous layer as a result of the activation of porphyrin complexes by TiO$_2$. The optical properties of porphyrin could change due to a special porphyrin complex containing both hydrophobic and hydrophilic parts as well as due to labile chlorine atoms associated with the central tin atom.

Sensor response to Valine is shown in Figures 4, 5. It was found that absorption of TiO$_2$-porphyrin decreased with increase of Valine concentration (Figure 4).

It was found that initially, the porphyrin showed low sensitivity to Valine (Figure 5, inserted plot). The significant changes of PL intensities and peak positions observed after adsorption of Valine on TiO$_2$-porphyrin surface (Figure 5). Adsorption of Valine led to a quenching and a blue-shift of the porphyrin emission band. At the same time, an increase of the intensity of TiO$_2$ emission was observed.

The obtained results point to the irreversible interaction between porphyrin and amino acid, resulted in the formation of new complexes between them and a reduction of optical activity of porphyrin.

The sensor signal was calculated using photoluminescence and absorption data S$_{lumin}$ (and S$_{ads}$):

$$S = \frac{S_0 - S_{Val}}{S_0}, \qquad (1)$$

where S$_0$ and S$_{Val}$ are PL (and absorption) signals of TiO$_2$-porphyrin nanostructure related to porhyrin emission and absorption, measured before and after Valine adsorption, respectively. The sensitivity of the sensor was obtained as the ration of the sensor response S$_{lumin}$ (and S$_{ads}$) due to (1) to the corresponding concentration of amino acid[26] C.

The sensitivity of the sensor vs Valine concentration is plotted in Figure 6. The obtained TiO$_2$ based sensor coated by porphyrin can detect Valine in the range of 0.04 to 0.16 mg/ml.

4. CONCLUSIONS

The TiO$_2$ and porphyrin form stable complex, proofed by the changes of absorption and PL of the porphyrin (IR shift) after deposition on TiO$_2$, matching to TiO$_2$-porphyrin interaction. TiO$_2$ nanostructure coated by porphyrin showed good properties for Valine detection. The irreversible interaction between TiO$_2$-porphyrin complex and Valine was confirmed by PL and absorption quenching after Valine adsorption and UV shift of PL peak position. The obtained results provide a basis for perspective applications of TiO$_2$-porphyrin nanostructures for effective detection of Valine.
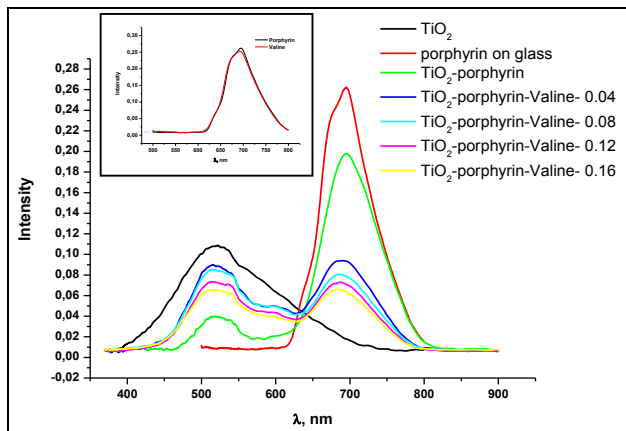
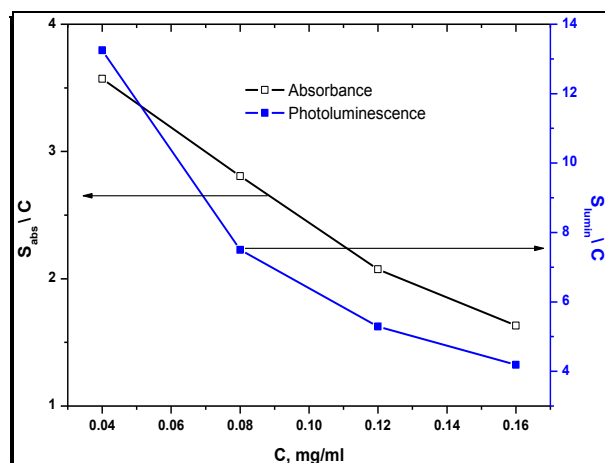Fig.5. **PL spectra of the TiO$_2$-porphyrin sensor measured at**

Fig.6. **Response of sensor for different concentrations different concentrations of Valine of Valine**

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Abstract

A novel optical sensor based on TiO$_2$ nanoparticles for Valine (one of the twenty standard amino acids within proteins) detection has been developed. In the presented work, commercial TiO$_2$ nanoparticles (Sigma Aldrich, particle size 32 nm) were used as sensor templates. The sensitive layer was formed by a porphyrin coating on a TiO$_2$ nanostructured surface. As a result, an amorphous layer between the TiO$_2$ nanostructure and porphyrin was formed. Photoluminescence (PL) spectra were measured in the range of 370-900 nm before and after porphyrin application. Porphyrin adsorption led to a decrease of the main TiO$_2$ peak at 510 nm and the emergence of an additional peak of high intensity at 700 nm. Absorption spectra (optical density vs. wavelenght, measured from 300 to 800 nm) also showed great changes; absorption edge shift and additional peaks appearing. Adsorption of amino acid resulted in a decrease of the intensity of the PL peak due to porphyrin and an increase of intensity of the TiO$_2$ main PL peak. The interaction between the sensor surface and the amino acid leads to the formation of new complexes on the surface and results in a reduction of the optical activity of porphyrin. Sensitivity of the sensor with respect to different concentrations of Valine was calculated. The developed sensor can determine the consentration of Valine in the range of 0.04 to 0.16 mg/ml.

Key words Titanium dioxide, nanoparticles, optical sensor, porphyrin, amino acid

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НАНОСТРУКТУРА TiO$_2$-ПОРФИРИН ДЛЯ ОПРЕДЕЛЕНИЯ АМИНОКИСЛОТЫ

Резюме

Разработан новый оптический датчик, основанный на наночастицах TiO$_2$ для обнаружения валина (одна из двадцати стандартных аминокислот среди белков). В представленной работе наночастицы коммерческого TiO$_2$ (Sigma Aldrich, размер частиц 32 нм) использовались как образцы датчика. Чувствительный слой был сформирован порфириновым покрытием на наноструктурированную поверхность TiO$_2$. В результате, между TiO$_2$ наноструктурой и порфирином был сформирован аморфный слой. Спектр фотолюминесценции (ФЛ) был измерен в диапазоне 370-900 нм до и после нанесения порфирина. Адсорбция порфирина приводила к уменьшению главного пика TiO$_2$ при 510 нм и появлению дополнительного пика большой интенсивности при 700 нм. Спектры поглощения (зависимость оптической плотности от длины волны измеренная в интервале от 300 до 800 нм) также проявляют большие изменения; сдвиг края поглощения и появление дополнительных пиков. Адсорбция аминокислоты проявилась уменьшением интенсивности пика ФЛ обусловленная порфирином и увеличением интенсивности главного пика ФЛ TiO$_2$. Взаимодействие между поверхностью датчика и аминокислотой ведет к формированию новых комплексов на поверхности и является результатом уменьшением оптической активности порфирина. Была рассчитана чувствительность датчика относитель-

но различных концентраций валина. Разработанный датчик может определять концентрацию валина в диапазоне 0.04 - 0.16 мг/мл.

Ключевые слова: Диоксид титана, наночастицы, оптический датчик, порфирин, аминокислота

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НАНОСТРУКТУРА TiO$_2$-ПОРФІРИН ДЛЯ ВИЗНАЧЕННЯ АМІНОКИСЛОТИ

Резюме

Розроблено новий оптичний датчик, заснований на наночастинках TiO$_2$ для виявлення валіну (одна з двадцяти стандартних амінокислот серед білків). У представленій роботі наночастинки комерційного TiO$_2$ (Sigma Aldrich, розмір частинок 32 нм) використовувалися як зразки датчика. Чутливий шар було сформовано порфіриновим покриттям на нано- структуровану поверхню TiO$_2$. У результаті, між наноструктурою TiO$_2$ і порфірином було сформовано аморфний шар. Спектр фотолюмінесценції (ФЛ) вимірювався у діапазоні 370-900 нм до і після нанесення порфірина. Адсорбція порфірина призводила до зменшення головного піка TiO$_2$ при 510 нм і появи додаткового піка великої інтенсивності при 700 нм. Спектри поглинання (залежність оптичної щільності від довжини хвилі вимірювана в інтервалі від 300 до 800 нм) також виявляють великі зміни; зсув краю поглинання і появу додаткових піків. Адсорбція амінокислоти проявилася зменшенням інтенсивності піка ФЛ обумовлена порфірином і збільшенням інтенсивності головного піка ФЛ TiO$_2$. Взаємодія між поверхнею датчика та амінокислотою веде до формування нових комплексів на поверхні і має результатом зменшенням оптичної активності порфірину. Було розраховано чутливість датчика щодо різних концентрацій валіну. Розроблений датчик може визначати концентрацію валіну в діапазоні 0.04–0.16 мг/мол.

Ключові слова: Діоксид титану, наночастинки, оптичний датчик, порфірин, амінокислота